

Effects of Fluorine Atoms Amount and Fluorinated Acrylic Chain Length Chemically Attached to Hydroxyl Groups on the Hydrophobic Properties of Cotton Fabrics

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Abstract

A two-step chemical treatment of cotton fabric was performed with the attachment of fluorine moieties on hydroxyl group sites. The hydroxyl groups of cotton were initially acrylated by 2-isocyanethyl methacrylate. Acrylic monomers, containing four to twelve fluorine atoms, had been used to build polymeric chain directly on the surface of cotton by means of radical polymerization. Fabrics became 10-20% stiffer and microscopy showed a clear change of the cotton surface after the treatment. Coated samples of cotton had shown hydrophobic property with a highest contact angle of 128 degrees. It was found that the increase of molecular weight of fluorinated polyacrylate on the cotton surface lowered contact angle value. The best results in hydrophobicity had been obtained at the molar stoichiometric ratio between the number of hydroxyl groups of cotton and the amount of fluorinated monomer added. This developed method allowed for the direct radical polymerization on cotton fabric, providing good hydrophobic properties with the formation of fluorinated polyacrylate of different molecular weights.

Keywords: Hydrophobicity; Fluorinated coating; Radical polymerization; Fluorinated acrylates; Contact angle

Introduction

Cotton is a popular natural fiber for making comfortable apparel. However, cotton absorbs water due to its abundance of hydroxyl (–OH) groups, thus reducing its mechanical and thermoinsulation properties. Various methods have been used to obtain water repellency on cotton surfaces, such as coating with water-resistant materials like nanoparticles [1] paraffin wax [2] silicon resin [3] or fluorocarbon [4]. Coating via the chemical transformation of –OH group is the widely known method for enhancing water resistance of cotton fibers and fabrics. Chemical transformation of –OH groups has some disadvantages such as long reaction time, involvement of non-ecologically friendly chemicals or solvents, necessity of special reactors as compared to methods using simple wetting in emulsions or coating with polymers. However, chemical transformation of –OH groups of cotton provides a better control of the process and higher washing durability as compared to physical coating due to formations of chemical bonds between cotton surfaces and water-repellent agents. These water-repellent agents could be in the forms of single molecules, oligomers or polymers. Among known chemical methods, fluorine is the best element to be used to lower the surface free energy and make fabric hydrophobic [5,6]. Fluorine has a small radius and a high electronegativity, thus the covalent bond between fluorine and carbon is extremely stable. When fluorine is replaced by other elements such as H and C, in the order $-CF_3 < CF_2H < CF_2 < CH_3 < CH_2$, the surface free energy is increased. Therefore, at the present, the impregnation of textiles with fluorine containing polymer dispersions or solutions is the most applied technique [7]. The increase in the number of fluorine-carbon bonds in the functional group attached to cotton will result in lower surface energy and higher hydrophobicity [8]. However, it has limitations such as poor adhesion between the substrate and the fluorine-based hydrophobic coating due to the weak van der Waals force [9]. Hence, chemical attachment of fluorinated water repellent agent to a cotton surface is a prominent direction that allows for the creation of a stable hydrophobic surface. And fluorinated acrylates are the most common and convenient substances for hydrophobic coating [10]. Currently, chemical attachment of fluorine-containing agents to the cotton surface is being developed by

three different approaches. The first direction encompasses synthesis of various small molecules and polymers, which can react with –OH groups on the cotton surface. For example, Schondelmaier [11] had synthesized fluoroalkylsilane molecules with hydrophobic and hydrophilic parts, where the hydrophilic parts of the molecules react with –OH groups. An interesting acrylate-based fluorinated agent with six fluorine atoms monomer had also been synthesized by Yang [12]. The chemistry was realized in three steps and the final product has shown good water-repellent properties when applied on cotton fabrics. A clear benefit for industrial application is that this treatment of cotton can be realized within a short time. A good water-repellent property was obtained at the usage of diblock copolymer in which one part provides water-repellent property whereas another part bearing alkoxy silane or epoxyde group can be chemically attached to the OH group on the cotton surface [13]. The second approach consists of the development of one-step methods of the fluorinated coating of cotton fabric. For example, a one-step process of attachment of a fluorinated monomer to the surface of cotton had been shown by Cai [14] who used irradiation-induced graft polymerization for direct deposition of nonafluorohexyl-1-acrylate on the cotton surface. In another example, direct fluorination of cotton had been done using elemental fluorine [15]. Another direction consists in the modification of –OH groups on a cotton surface into other functional groups, which are to be used for attachment of different water repellent agent, can be done. For example, Yu [16] successfully oxidized the primary –OH groups of cotton by laccase/TEMPO treatment, obtaining aldehyde instead of –OH group, which were then used for the reaction with octadecylamine.

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Results

Characterization of obtained cotton samples

Successful deposition of hydrophobic coatings on the cotton sample was evaluated by comparing the FTIR spectra of coated cotton samples with the untreated sample. FTIR spectra of untreated and treated cotton fabrics are presented in Figure 1. The peak near 1730 cm^{-1} was attributed to the stretching vibration of the carbonyl group of acrylates monomers attached to the fabric. Because of a small amount of fluorinated polymer on the cotton surface and overlapping of peaks typical for $-\text{CF}_3$ group, there are no further clearly distinguishable differences between treated and untreated cotton. All presented SEM photos were taken using cotton samples obtained using the ratio of 1:1 of $-\text{OH}$ groups to fluorinated monomer because this ratio has shown higher values of a contact angle. The SEM image of the untreated cotton sample (Figure 2) had shown a smooth surface with characteristic parallel ridges. The primary wall structure was also obvious from the SEM image. The SEM image of the treated cotton samples did not exhibit the fine fibril structure that was present in the starting material. Thin laminates and patches with irregular edges were observed. These may have resulted from the rearrangement of surface cellulose due to the formation of a thin fluorinated layer in the primary wall. To characterize the chemical change of cotton after the treatment, quantitative EDX analysis of elements were made using the same samples chosen for SEM. The fluorine content found on the surface of the differently treated cotton is presented in Table 1. As seen in Table 1, the amount of fluorine on the surface of cotton had increased from 0.46% for sample coated with TFPA up to 0.90% for the sample coated with DFHA. Theoretically, if all $-\text{OH}$ groups were modified and covered by fluorinated monomers, the amount of fluorine in the treated cotton should be 6.2% wt. for TFPA and up to 12.9% wt. for DFHA. Therefore, the degree of fluorination of acrylated $-\text{OH}$ group can be approximately estimated to be 7%.

Water repellent property

The average contact angles of the differently treated cotton fabrics and different ratios of the number of double bonds on the cotton surface to the amount of fluorinated monomer are presented in Table 2. It should be noted that the untreated contact angle of cotton fabrics could not be measured because the water droplet was quickly absorbed into the fabric within 4 seconds. This is due to the high hydrophilic properties of cotton. The intermediate product, where acrylic groups are attached on the surface of cotton, has a contact angle

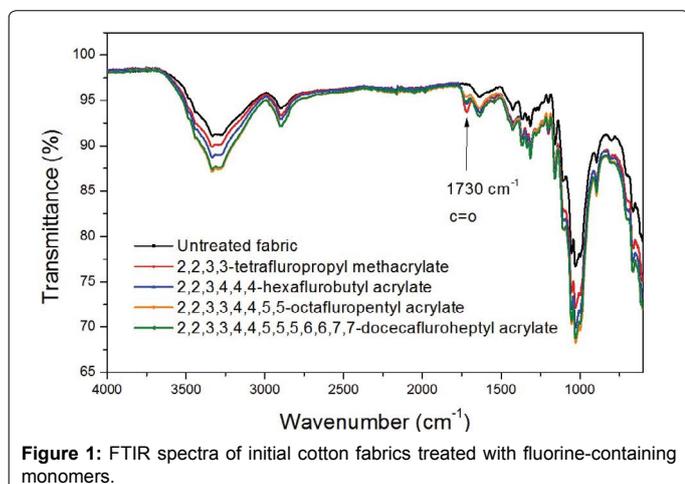


Figure 1: FTIR spectra of initial cotton fabrics treated with fluorine-containing monomers.

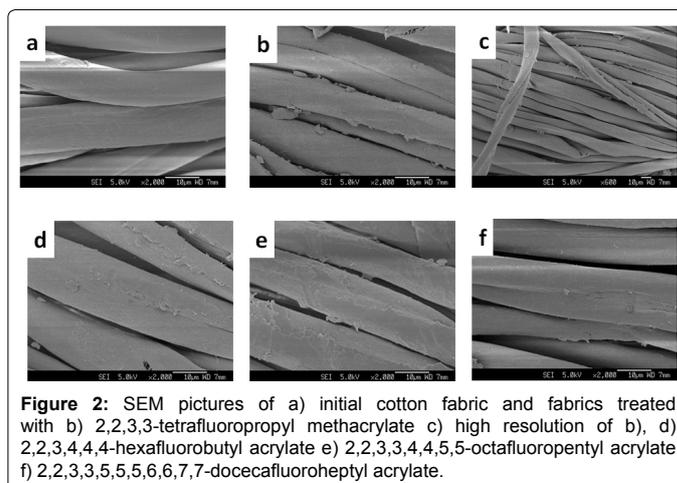


Figure 2: SEM pictures of a) initial cotton fabric and fabrics treated with b) 2,2,3,3-tetrafluoropropyl methacrylate c) high resolution of b), d) 2,2,3,4,4,4-hexafluorobutyl acrylate e) 2,2,3,3,4,4,5,5-octafluoropentyl acrylate f) 2,2,3,3,5,5,6,6,7,7-docecafluoroheptyl acrylate.

Type of fluorinated monomer	Amount of fluorine % mass
TFPA	0.46
HFBA	0.78
OFPA	0.79
DFHA	0.90

Table 1: Amount of fluorine in the treated cotton samples.

Type of fluorinated monomer	Average contact angle (°) Ratio of Double bond on surface: Fluorinated monomer		
	1:1	1:5	1:10
TFPA	107.7	103.6	89.5
HFBA	113.2	105.5	101.9
OFPA	123.6	122.6	107.8
DFHA	128.2	122.8	115.4

Table 2: Contact angle values for differently treated cotton fabrics.

of 98.3° , which confirmed that there had been a modification of the $-\text{OH}$ groups. However, it was still lower than that of fluorine-treated cotton fabric, due to the absence of hydrophobic tail group (C-F). It can be seen that the cotton treated with longer chains of fluorinated monomers were more hydrophobic as compared to that with shorter chains. This might be due to nonpolar carbon-fluorine (C-F) chains decreasing the overall polarity of the molecule. In addition, longer carbon chains increase the hydrophobicity of the fabric surface, resulting in greater contact angle. As a general trend, the surface energy of cotton fabric decreases as the carbon chain lengths and amount of fluorine atoms increase. In addition, the introduction of long fluorine chains creates rougher surfaces on the cotton fibers, and this plays a positive role in hydrophobicity by decreasing the surface free energy as well. The increase in the molar ratio of the fluorinated monomer has an interesting influence on hydrophobicity. The higher amount of fluorinated monomers we add, the lower hydrophobicity of fabric we obtain. A large molecular chain on the cotton surface could create active sites between the fluorinated monomer chain and the substrate. In particular, a 1:10 molar ratio could create a multiple-linking system, as was mentioned by Colleoni [19], in which the influence of the fluorine chains steric hindrance is balanced by an increase in the number of links between the coating and the cotton fabric sample substrate. Long macromolecule with fluorine atoms on the surface of cotton is also more inclined to make some compact structures such as crystals, coils or globules where fluorine atoms could be hidden and become unavailable to impact hydrophobicity.

Mechanical properties of modified fluorinated cotton

It had been observed that there were slight differences between the mechanical properties of untreated and treated cotton. Young moduli of all samples were fluctuating within the same range. Untreated cotton had the lowest modulus (201.0 N/mm²) while the sample with TFPA treatment had shown the highest (328.4 N/mm²). The treated samples showed decreasing modulus with increasing number of fluorine up to 272.0 N/mm² for DFHA treated sample. Measurements of the mechanical properties of the untreated and treated cotton fabrics are given in Figure 3. From the graphs, it could be seen that samples of fabric became stiffer after treatment, with the maximum stress value being 60% higher for fabrics treated with TFPA as compared to the untreated fabric. It had also been observed that the treated cotton fabrics had a loss in strain and elongation at break values almost twice as compared to untreated fabric. It could be related to the adhesion between cotton fibers. In addition, the formation of polymeric bridges between cotton fibers due to polymerization could also be possible. This phenomenon had also been mentioned by Maity [15]. An interesting observation was the value of stress connected with the amount of fluorine atoms in monomers used for treatment. Monomers with higher fluorine content used for treatment leads to lower stiffness and stress value of the fabric. It could likely be related to the crystallinity of polymer formed on the cotton surface. Lower fluorine content and shorter length of monomer tends to result in higher crystallinity at the same degree of polymerization.

Conclusion

A thin coating with fluorinated polymer chemically attached to the cotton fabric was shown. The amount of fluorine in the final fabric is lower than 1%wt, which is lower when compared to other similar treatments, and yet the fabric is showing good water-repellent property. The cotton that was treated with 2,2,3,3,4,4,5,5,5,6,6,7,7-docecafluoroheptyl acrylate obtained the highest contact angle of 128°. Although modification of cotton can allow for better hydrophobicity, mechanical properties need not be sacrificed. It had been found that by making the length of the attached polymer longer, not only will the mechanical properties not be affected significantly but at the same time it gives the best result in terms of hydrophilicity. It can also be said that there will be a smaller environmental impact as less fluorine will go into the environment during the utilization steps of fluorinated cotton fabric

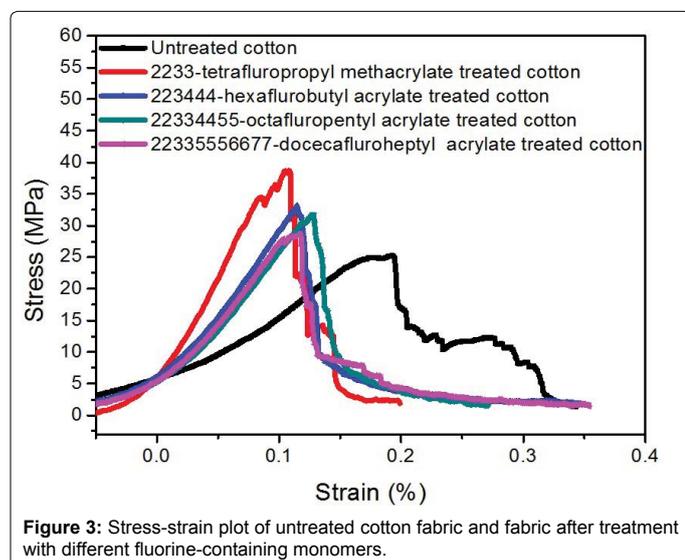


Figure 3: Stress-strain plot of untreated cotton fabric and fabric after treatment with different fluorine-containing monomers.

as well as the usage. The developed method opens a wide area allowing for coating with layers, using a combination of acrylic monomers for coating, making different lengths of hydrophobic polymers on the cotton, and creating other properties, for example antimicrobial, of fabric.

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